Linear Furano Compounds: Synthesis of 7H-Furo[3,2-g][1]benzopyran-7-ones

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Synopsis. A novel synthesis of 2-aroyl-7*H*-furo[3,2-g]-[1]benzopyran-7-ones, viz., IIIa and b was reported by condensing 2-benzoyl- and 2-(p-methoxybenzoyl)-3-methyl-5acetyl-6-hydroxybenzofuran (Ia and b) with C₆H₅CH₂-COONa-Ac2O. VI was synthesised by condensing 7-hydroxy-6-benzoyl-4,8-dimethylcoumarin (IV) with V. The basic ethers (IIId-j) were synthesised by demethylating IIIb with pyridinehydrochloride and resultant IIIc was condensed with N,N-dialkyl-2-haloalkanamine hydrochlorides. UV, IR, ¹H NMR, and Mass spectral data are also given.

2-(p-Hydroxybenzoyl)benzofuran, 1) 2-(p-hydroxybenzoyl)-3-ethylbenzofuran,1) unsubstituted -7H-furo-[3,2-g][1]benzopyran-7-one, psoralene²⁾ and substituted -7H-furo[3,2-g][1]benzopyran-7-ones,³⁾ viz., 3-[p-(2-dimethylamino-1-methylethoxy)phenyl]-, 3-[p-[2-(diethylamino)ethoxy]phenyl]-, and 3-[p-(3-piperidinopropoxy)phenyl]-substituted 2-phenyl-9-methyl-7H-furo-[3,2-g][1]benzopyran-7-ones have shown antifertility activity. The antifertility activity of these compounds is attributed to the presence of p-hydroxybenzoyl group and p-[2-(dialkylamino)alkoxy]-benzoyl and p-[2-(dialkylamino)alkoxy]phenyl groups.

Results and Discussion

Based on the above reports, 2-[p-[2-(dialkylamino)alkoxy]-benzoyl]-7H-furo[3,2-g][1]benzopyran-7-ones-(**IIId**—i) were synthesised by developing the 2-pyrone ring on 2-benzoyl-and 2-(p-methoxybenzoyl)-3-methyl-5-acetyl-6-hydroxybenzofuran(Ia and b). 2-Benzoyl- and 2-(p-methoxybenzoyl)- 3,5-dimethyl-6-phenyl-7*H*-furo-[3,2-g][1]benzopyran-7-ones(IIIa and b) were synthesised to study structure and antifertility activity relationships. † IIIa and b were synthesised by refluxing a mixture of **Ia** and **b**, C₆H₅CH₂COONa and Ac₂O at 180°C for 20—22 h in 70 and 80% yields. This method provides a convenient synthetic route for the synthesis of -7Hfuro[3,2-g][1]benzopyran-7-ones IIIa and b (Scheme 1). IIIa was also synthesised in 50% yield by refluxing a mixture of Ia with phenylacetyl chloride and freshly baked K₂CO₃ in dry acetone. IIIb was demethylated with pyridine hydrochloride to give 2-(p-hydroxybenzoyl)-3,5-dimethyl-6-phenyl-7*H*-furo[3,2-*g*][1]benzopyran-7one(IIIc). IIIc was then, condensed with a number of N,Ndialkyl-2-haloalkanamine hydrochlorides in 1:1 mol ratio, and freshly baked K₂CO₃ in dry acetone for 20— 22 h to give **IIId**—j (Scheme 1) in 60 to 70% yield.

2-Benzoyl-3-phenyl-5, 9-dimethyl-7H-furo[3, 2-g][1]benzopyran-7-one(VI) (Scheme 2) was synthesised by condensing 7-hydroxy-6-benzoyl-4,8-dimethylcoumarin-(IV) with ω -bromoacetophenone (V) in 1:1 mol ratio freshly baked K₂CO₃ in dry acetone for 12 h (Scheme 2). The spectral data is summarized in Table 1. All the

2-[p-[2-(Dialkylamino)alkoxy]benzoyl] furo[3,2-g]-[1]benzopyran-7-ones

where R

Scheme 1.

compounds exhibited two distinct UV absorption maxima in the region λ 288—292 nm and 355—358 nm. The IR spectra of these compounds have shown a lactone carbonyl group absorption at 1705—1720 cm⁻¹ and an aroyl carbonyl group absorption at 1620— 1630 cm⁻¹. The linear nature of the compounds IIIa, b, d, and g is confirmed by the observation of two singlets for the protons C₄ and C₉ in the ¹H NMR spectra of these compounds in the region δ 7.88—7.83 and δ 7.48—6.9 respectively. Further, the ¹H NMR spectrum, of compound **IIId** displayed two triplets at δ 4.1 and 2.8 respectively for

[†]Antifertility activity is reported elsewhere.

Table 1. Characterization Data of IIIa-i and VI

Compd ^a	a-c) Mp θ _m /°C	Formula	$UV^{d)}$ $\lambda_{max}/nm\ (\logarepsilon)$ -	IR (KBr) -C=O/cm ⁻¹		¹ H NMR (CDCl ₃) δ (I in Hz)
				Lactone	Aroyl	(J 112)
IIIa	208	$C_{26}H_{18}O_4$,	290(4.23), 355(4.23)	1705	1625	7.98(m, 2H, 2',6'-H), 7.83(s, 1H, 4-H), 7.35(m, 8H, 6-C ₆ H ₅ -H, 3',4',5'-H), 7.48(s, 1H, 9-H), 2.62(s, 3H, 5-CH ₃), 2.38(s, 3H, 3-CH ₃)
IIIb	270	C ₂₇ H ₂₀ O ₅	290(4.48), 355(4.50)	1705	1625	8.25(d, 2H, <i>J</i> =9, 2',6'-H), 7.88(s, 1H, 4-H), 7.32(m, 5H, 6-C ₆ H ₅ -H), 6.95(d, 2H, <i>J</i> =9, 3',4'-H), 7.18(s, 1H, 9-H), 2.65(s, 3H, 5-CH ₃), 2.38(s, 3H, 3-CH ₃), 3.95(s, 3H, 4'-OCH ₃ -H)
IIIc	286	$C_{26}H_{18}O_5$	290(4.87), 358(4.92)	1680	1625	
IIId	110	C ₃₀ H ₂₇ NO ₅	292(4.28), 358(4.36)	1715	1622	7.83(d, 2H, J =9, 2′,6′-H), 7.63(s, 1H, 4-H), 7.13(m, 5H, 6-C ₆ H ₅ -H), 6.75(d, 2H, J =9, 3′,5′-H), 7.0(s, 1H, 9-H), 4.1(t, 2H, J =6, -O-CH ₂ -), 2.8(t, 2H, J =6, -CH ₂ -N $\stackrel{<}{\sim}$), 2.61(s, 3H, 5-CH ₃), 2.35(s, 6H, -N-(CH ₃) ₂), 2.30(s, 3H, 3-CH ₃)
IIIe	103	$C_{31}H_{29}NO_5$	292(4.74), 355(4.77)	1710	1620	_
IIIf	95—97	$C_{32}H_{31}NO_5$	290(4.30), 355(4.36)	1710	1620	-
IIIg	101	C ₃₂ H ₂₉ NO ₅	290(4.30), 358(4.36)	1715	1625	7.9(d, 2H, J =9, 2',6'-H), 7.64(s, 1H, 4-H), 7.1(m, 5H, 6-C ₆ H ₅ -H), 6.71(d, 2H, J =9, 3',5'-H), 6.9(s, 1H, 9-H), 4.1(t, 2H, J =6, -O-CH ₂ -), 2.8—2.74(m, 5H, -CH ₂ -N \subset , 5-CH ₃), 2.7—2.2(m, 7H, -N-(CH ₂) ₂),
						1.8(m, 4H, -N CH ₂
IIIh	108	C33H31NO5	290(4.42), 355(4.47)	1720	1625	— C⊓2
IIIi	98	$C_{33}H_{31}NO_5$ $C_{34}H_{33}NO_5$	290(4.42), 353(4.47) 292(4.44), 258(4.53)	1720	1625	-
IIIj	105	C ₃₂ H ₂₉ NO ₆	288(4.33), 355(4.91)	1720	1625	<u>_</u>
VI	190—192	$C_{26}H_{18}O_4$	290(4.45), 358(4.50)	1715	1625	_

a) Satisfactory elemental analysis obtained for all the compounds. b) Compounds **IIIa**—c were obtained in 80, 70, and 90% yield, respectively. Compounds **IIId**—j and VI were obtained in 60—65 and 50% yield respectively. c) All the compounds were crystallized from dioxane. d) The UV spectra of compounds **IIIa**—f, h, j, and **IIIg**, i were recorded in chloroform and methanol respectively.

the protons of $-O-CH_2-$ and $-CH_2-N \le$ of (alkylamino)-alkoxyl group, whereas in the case of compound **IIIg**, the protons of $-O-CH_2-$ have displayed a triplet and the protons of $-CH_2-N \le$ and C_5-CH_3 groups have been observed as a multiplet in the region δ 2.80—2.74. The

Scheme 2.

mass spectra of compounds **IIIa**, **b**, and **VI** showed molecular ions M+394(8.1%), M+424(100%), and M+394(1.5%) respectively.

Experimental

Melting points of all the compounds **IIIa—j** and **VI** were taken in open capillaries and are uncorrected. UV absorption spectra were taken on Shimadzu UV 240 Ultraviolet Visible spectrophotometer, IR spectra were recorded on Perkin-Elemer spectrophotometer, and ¹H NMR spectra on a Varian EM 390-90 MHz NMR spectrophotometer with TMS as an internal standard.

The procedures given below are typical of the general procedure employed.

2-(*p*-Methoxybenzoyl)-3,5-dimethyl-6-phenyl-7*H*-furo[3,2-*g*]-[1]benzopyran-7-one (IIIb). Five ml of Ac_2O was added to an intimate mixture of Ib (16.20 g; 0.05 mol) and sodium phenylacetate (15.8 g; 0.1 mol), and the mixture was refluxed in an oil bath at 180° for 24 h. The reaction mixture was cooled and poured onto a crushed ice (5g). The solid obtained was recrystallized from dioxane. Yield: 14 g (70%), Found: C, 76.37; H, 4.65%. Calcd for $C_{27}H_{20}O_5$: C, 76.41; H, 4.71%

2-(p-Hydroxybenzoyl)-3,5-dimethyl-6-phenyl-7H-furo[3,2-g]-[1]benzopyran-7-one (IIIc). IIIb (12. 72 g; 0.03 mol) was added to the freshly distilled pyridine hydrochloride (34.5 g;

0.3 mol), and the mixture was refluxed for 30 min. The cooled reaction mixture was poured into water, and the solid filtered off, washed with water, dried, and crystallized from dioxane. Yield: $10.8\,\mathrm{g}(90\%)$. Found: C, 76.02; H, 4.31%. Calcd for $C_{26}H_{18}O_5$: C, 76.09; H, 4.39%.

2-[p-(2-Pyrrolidinylethoxy)benzoyl]-3,5-dimethyl-6-phenyl-7H-furo[3,2-g][1]benzopyran-7-one (IIIg). IIIc (1.23 g; 0.003 mol) was dissolved in acetone (400 ml) and to this 1-(2-chloroethyl)pyrrolidine hydrochloride (0.510 g; 0.003 mol) and freshly baked K₂CO₃ (3g) were added, and the mixture was refluxed for 22 h. The inorganic impurities were filtered off and the residue was poured into ice cold water, filtered, washed with water, and crystallized from aqueous dioxane as colorless flakes. Yield: 0.950 g(65%). Found: C, 75.69; H, 5.65; N, 2.69%. Calcd for C₃₂H₂₉NO₅: C, 75.73; H, 5.71; N, 2.76%

2-Benzoyl-3-phenyl-5,9-dimethyl-7*H*-furo[3,2-g**I**1]benzopyran-7-one(VI). A mixture of **IV** (0.294 g; 0.001 mol) and ω-bromo-acetophenone (0.199 g; 0.001 mol) in acetone (300 ml) was refluxed for 12 h cooled, and the inorganic impurities were filtered off. Acetone was recovered by distillation and the resi-

due was poured into ice cold water. The solid obtained was recrystallized from dioxane as shining needles. Yield: 0.908 g(50%). Found: C, 79.10, H, 4.49%. Calcd for $C_{26}H_{18}O_4$: C, 79.18; H, 4.56%.

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